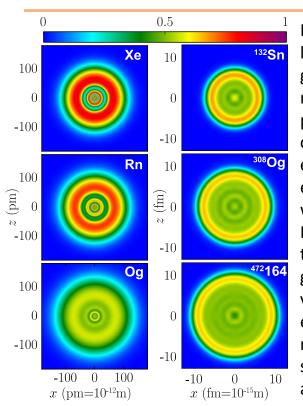


Oganesson is an oddball among atoms and nuclei



Objectives

- Oganesson (Og, Z118) is a recent addition to the periodic table of the elements and the chart of nuclides.
- Using state-of-the art atomic and nuclear models, and advanced computational tools, we study the electronic and nucleonic shell structure of superheavy elements.



Left: Electron localization for noblegas elements xenon, radon, and oganesson predicted in relativistic calculations. The outer electron shells of Og are expected to show weaker shell effects. **Right: A similar** transition to uniformgas behavior in the valence region is also expected for the neutron localization of superheavy nuclei ³⁰⁸Og and ⁴⁷²164 (right).

Impact

- Og is a rather unusual addition to the periodic table and to the chart of nuclides: it breaks our textbook picture of electronic and nucleonic shells.
- Og is expected to show uniform-gas-like behavior in the valence region both for electrons and neutrons. Consequently, Og is expected to be a metal at room temperature.

Accomplishments

- Publication: P. Jerabek, B. Schuetrumpf, P. Schwerdtfeger, and W. Nazarewicz, <u>Phys. Rev.</u> <u>Lett. 120, 053001 (2018).</u>
- Highlighted as Editors' suggestion.
- Featured in Physics (phys.aps.org) as Viewpoint: <u>Heaviest Element Has Unusual Shell Structure</u>
- Featured on the cover of Phys. Rev. Lett. 120(5) issue.
- Featured by <u>Physics Today</u>, <u>Nature</u>, <u>Physics World</u>, <u>Science News</u>, <u>Gizmodo</u>, <u>Phys.org</u>, <u>Chemistry</u> <u>World</u>, and many international news outlets.